

Code No. 3508

FACULTY OF SCIENCE
M.Sc. I - Semester Examination, December 2014

Subject : Physical Chemistry
Paper – III

Time: 3 hours

Max. Marks: 80

Note: Answer all questions from Part - A and Part - B.
Each question carries 4 marks in Part - A and 12 marks in Part - B.

PART – A (4 x 8 = 32 Marks)
(Short Answer Type)

- 1 a) State third law of thermodynamics.
b) Write the physical significance of Helmholtz (A) and Gibbs (G) free energies
- 2 a) Define the terms i) decomposition potential ii) Concentration over potential
b) Briefly explain the elementary treatment of Bjerrums theory of ion association.
- 3 a) Explain the plancks concept quantization and give the expression for average energy of an oscillator.
b) Write the interpretation of wave function ψ and ψ^2 .
- 4 a) What is Lindmanns theory?
b) Explain the collision theory of reaction rates.

PART – B (4 x 12 = 48 Marks)
(Essay Answer Type)

- 5 a) Derive the expression for entropy change in mixing of ideal gases.
b) What are Maxwell relations? Show that $\left(\frac{\partial H}{\partial V}\right)_s = -T\left(\frac{\partial P}{\partial H}\right)_v$.
- OR**
- c) Write the Gibbs-Helmholtz equation and explain the dependence of G on T and P.
d) ΔG for a reaction at 300 K is – 16k.cal. ΔH for reaction is – 10k.cal. Calculate the entropy change (ΔS). What will be the ΔG at 330K.
- 6 a) Derive the Nernst equation and explain terms in it.
b) Write the Debye-Huckel-Onsager equation and explain its limitations and validity.
- OR**
- c) Explain the principle in potentiometric redox titration and cell representation with an example.
d) Explain the Debye-Huckel theory of electrolytic solution and Debye-Huckel limiting law.
- 7 a) State and explain postulates of quantum mechanics.
b) Show that the eigen value of Hermitian operator is real.
- OR**
- c) Derive the time independent Schrodinger wave equation from time dependant Schrodinger wave equation.
d) What are commuting operators? Show that the commutator of $[d/dx, x]$ is one.
- 8 a) Write the Hammett equation and explain the reasons for deviations from Hammett's correlations.
b) Explain the reactivity selectivity principle and isoselectivity rule.
- OR**
- c) What is Tafts four parameter equation? Explain the terms in it and its applicability.
d) Write notes on i) Activation parameter ii) Swain-Scott equation
